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㉗ Method for producing formed products of high molecular compounds.

㉘ A formed product of a high molecular compound having a glass transition temperature of not more than about 60°C, which is substantially free of halogenated solvents harmful to the human body, is produced by mixing the high molecular compound, a halogenated solvent and an organic solvent other than the halogenated solvent which is compatible with the solvent and is less apt to dissolve the high molecular compound to prepare a solution and then maintaining the solution at a temperature around the boiling point of the halogenated solvent.

- 1 -

METHOD FOR PRODUCING FORMED PRODUCTS OF  
HIGH MOLECULAR COMPOUNDS.

This invention relates to a method for producing formed products of high molecular compounds having a glass transition temperature of not more than about 60°C. More particularly, the present invention is concerned with a method for producing formed products of high molecular compounds, which method comprises using a halogenated solvent to produce a formed product of a high molecular compound, which is substantially free of the halogenated solvent.

Halogenated solvents are well known as good solvents for dissolving high molecular compounds. Also, they are non-flammable or practically non-flammable as compared with other organic solvents. This provides advantages in production and contributes to their wide application.

However, halogenated solvents present hazards of liver disturbances, carcinogenicity and teratogenicity; thus, these substances are highly harmful to the human body, cause extremely increased tissue permeability, and are absorbable rapidly through any part of the human body.

In order to remove halogenated solvents having such characteristic properties after using halogenated solvents to produce the desired formed product, only

temperature of not more than about 60°C,

(b) a halogenated solvent and

(c) an organic solvent other than the halogenated solvent which is compatible with the halogenated solvent and is less apt to dissolve the high molecular compound, to prepare a solution of the high molecular compound, and then maintaining the solution at a temperature around the boiling point of the halogenated solvent so as to remove the halogenated solvent.

As the high molecular compound employable in the present invention, one which has a glass transition temperature of not more than about 60°C is employed, and one which has a glass transition temperature of not less than about 30°C is preferred.

As to glass transition temperature, reference is made to "Rheology and its application", Renpei Goto et al., page 22, 1967, Kyoritsu Publishers, Japan.

The molecular weight of the high molecular compound is not limited, but is preferably about 5000 to 30000.

Examples of the high molecular compound, which is usable in this invention, include for example polymers which are soluble in halogenated solvents, such as polyester type compounds (e.g. polylactic acid, copolymers of lactic acid and glycolic acid), and waxes such as carnauba wax, paraffin and glycerides.

The halogenated solvents which are employable in this invention are preferably those having a boiling point of about 10°C to 100°C and having 1 to 6 halogen atoms and their examples include for instance halogenated methane (e.g. methylene chloride, chloroform, carbon tetrachloride, trichloromonofluoromethane, difluorodibromomethane), halogenated ethane (e.g. ethyl chloride, ethyl bromide, 1,1,2-trifluoro-1,2,2-trichloroethane, s-tetrachlorodifluoroethane, dibromodifluoroethane), ethylene chloride, ethylidene chloride, trichloroethylene, allyl

chloride, isopropyl chloride and butyl chloride. In particular, halogenated methane is preferred, and methylene chloride is more preferred. The amount of halogenated solvent used is that which dissolves the high molecular compound or more than that.

The organic solvent (hereinafter referred to as "present auxiliary agent for removal") other than the halogenated solvents, which is compatible with the solvents to be used in this invention and is less apt to dissolve the high molecular compounds to be employed in this invention, may be any type of solvent, provided it possesses the above-described properties.

The present auxiliary agent for removal is preferably one having a boiling point within about  $\pm 50^{\circ}\text{C}$  from the boiling point of the above-mentioned halogenated solvent, preferably within about  $\pm 20^{\circ}\text{C}$ .

The expression "compatible with the solvent" means substantially completely soluble in the said solvent.

The expression "less apt to dissolve the said high molecular compound" denotes that the present auxiliary agent for removal, alone, fails to dissolve the high molecular compound or has merely a solubility of not more than about 3 % (V/V).

Specific examples of the present auxiliary agent for removal include, for example, straight-chain, branched-chain or cyclic hydrocarbons of 5 to 8 carbon atoms, ketones, ethers, esters and alcohols, and their boiling points are preferably about 10 to  $100^{\circ}\text{C}$ .

Representative examples of the said hydrocarbons include, for example, n-pentane, isopentane, n-hexane, cyclohexane, cyclohexene, benzene, pentene and hexene; representative examples of the said ketones include for example acetone and methyl ethyl ketone; representative examples

of the said ethers are, for example, diethyl ether, methyl ethyl ether and tetrahydrofuran; representative examples of the said esters are, for example, ethyl acetate and methyl acetate, and representative examples of the said alcohols, for example are methyl alcohol and ethyl alcohol, respectively.

The present auxiliary agent for removal may be added after the high molecular compound is dissolved in the halogenated solvent, or, alternatively, may be mixed in advance with the solvent. Furthermore, it causes no inconvenience to add the present auxiliary agent for removal previously to the high molecular compound, etc., followed by addition of the solvent.

The amount of the present auxiliary agent employed in the present method for removal is not more than about 50 % (V/V) based on the quantity of the halogenated solvent, preferably about 5 to 30 % (V/V).

As the formed product obtained by using the solution of high molecular compound, there are mentioned, for example, film-like products, bar-shaped or block-shaped products, powders, film coating products on tablets or granules, microcapsules and microspheres.

The phrase "to maintain at a temperature around the boiling point of the halogenated solvent" as referred to in the process of this invention means a procedure of maintaining at a temperature within about  $\pm 20^{\circ}\text{C}$  of the boiling point of the halogenated solvent, more preferably, at a temperature within about  $\pm 10^{\circ}\text{C}$  thereof.

In the maintaining at the above-described temperature, it is preferable to conduct such a procedure under reduced pressure, for example, under reduced pressure of not more than about 10 Torr, more preferably not more than about 1 Torr.

The above-described procedure may be conducted in a

thermostatically-controlled vessel maintained under reduced pressure, fluidized vessel, moving bed or kiln.

Alternatively, the formed product may be maintained at the temperature in an air-tight sealed or closed  
5 system together with a substance (e.g., activated carbon, diatomaceous earth, etc.) capable of absorbing or adsorbing the halogenated solvent.

The length of time during which the above-described procedure is conducted is changed, depending upon the  
10 shape or form, thickness, surface area and treated amount of the formed product, the types and quantities of the solvent, the present auxiliary agent for removal and the pressure, and also varies according to the intended remaining amount, for example it is normally in the range  
15 of about 1 hour to 48 hours. The procedure continues till the formed product is substantially free of the halogenated solvent.

By following the above-described procedure, there is obtained a formed product of high molecular compounds  
20 which is substantially free of the halogenated solvent. The expression "substantially free of" as mentioned above denotes concentration of not more than 1 ppm or below the sensitivity of detection as measured by a given detection apparatus, for example, gas  
25 chromatography equipped with ECD (electron capture detector).

The method according to this invention can produce a formed product of the high molecular compound which is substantially free of halogenated solvent,  
30 and, therefore, the formed product is safe for humans. Furthermore, the present formed product brings less deformation with the lapse of time due to the decline of plastic deformation of the formed product per se in comparison with a formed product which retains a large  
35 amount of halogenated solvent. When the present

formed product is stored in a packing system, the present formed product does not cause deterioration or denaturation of the packing material due to the evaporation of the halogenated solvent. Furthermore, it does not require any special machinery and equipment for the removal of the halogenated solvent in the production of the formed product of the high molecular compound on an industrial scale. Therefore, the method of this invention can be advantageously employed in producing a formed product of the high molecular compound on an industrial scale.

The examples are described below to illustrate this invention more concretely.

Example 1

In a mixed solution comprising 5 ml of methylene chloride and 1 ml of n-pentane was dissolved 4 g of polylactic acid (average molecular weight: about 20000, glass transition temperature: about 48 to 53°C), and the mixture was formed into a film on a Teflon<sup>®</sup> plate, followed by air-drying for 24 hours. Subsequently, the formed film was placed under the conditions of 0.2 Torr and 42°C to remove the solvent. The remaining amount of methylene chloride as determined is shown in the following table.

25

| Conditions     | Time (h) | Remaining amount |
|----------------|----------|------------------|
| 42°C, 0.2 Torr | 0        | 18000 ppm        |
|                | 6        | 100 ppm          |
|                | 12       | Not detected     |
|                | 24       | Not detected     |
|                | 48       | Not detected     |
|                | 96       | Not detected     |

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The measurement of the remaining amount of the

solvent was carried out by gas chromatography equipped with ECD. The same shall apply in the Examples to be described in the following.

Example 2

5           In a mixed solution comprising 100 parts of carbon tetrachloride and 30 parts of cyclohexene was dissolved 20 parts of polylactic acid (average molecular weight: about 20000, glass transition temperature: about 48 to 53°C), and after the solution was filtered to  
10       remove foreign matter, etc., the solvent was evaporated by aeration. Subsequently, removal of the solvent was carried out at 0.1 Torr and at 80°C for 18 hours, and the molten polylactic acid was cooled and pulverized to give powdered polylactic acid. It showed a remaining amount  
15       of carbon tetrachloride of not more than 1 ppm.

Example 3

          In a hexagonal inclined revolving granulation machine was charged 2 kg of nonpareil spherical granules (32 to 28 mesh) produced from granulated sugar and corn  
20       starch, and the nonpareil granules were coated by a spray gun, with a solution of 100 g of n-paraffin (average molecular weight: about 500 to 600, glass transition temperature: about 50°C) in a mixed solution comprising 700 ml of chloroform and 300 ml of methanol at 50°C. The  
25       resulting spherical coated granules were mixed with 20 g of talc, and forced-air dried with warm air at 52°C for 24 hours. As a result, the granules showed a remaining amount of chloroform of not more than 1 ppm.

Example 4

30           In 2.5 ml of a 20 % aqueous gelatin solution warmed (at 60 to 70°C) in advance to convert to the liquid form was dissolved 200 mg of TAP-144 [an acetate of peptide represented by the formula (Pyr)Glu-His-Trp-Ser-Tyr-D-Leu-Leu-Arg-Pro-NH-CH<sub>2</sub>CH<sub>3</sub> which possesses an action  
35       similar to that of luteinizing hormone-releasing hormone

(LH-RH). The abbreviations used are in accordance with IUPAC-IUB Commission on Biochemical Nomenclature.

Optical isomers of the aminoacids refer to the L-isomer, unless otherwise specified, and the solution was added  
5 to 20 % of a lactic acid-glycolic acid copolymer [average molecular weight: 14,000, monomer ratio of lactic acid to glycolic acid: 75/25 (on a weight basis), glass  
transition temperature: about 40 to 45°C] in a mixed  
10 solution comprising 10 ml of methylene chloride and 3 ml of n-pentane, followed by ultrasonic treatment (20 KHz, 100 W for several minutes by use of an ultrasonic cell homogenizer, produced by Ohtake Seisakusho, Ltd., Japan) to prepare an adequately fine W/O emulsion. The emulsion was immediately cooled with ice to solidify the gelatin  
15 phase and poured into 1000 ml of 0.5 % polyvinyl alcohol (Gohsenol EG-40, produced by Nippon Synthetic Chemical Industry Co., Ltd., Japan) cooled in advance with ice, followed by dispersion in a homogenizer having 80  $\mu$ m punching metals at 3000 r.p.m. for 60 seconds to prepare  
20 a W/O/W emulsion. The emulsion was transferred rapidly into a rotary evaporator, whereby methylene chloride and n-pentane were removed under ice-cooling. After no bubbling was observed to take place, the residual solution was gradually warmed up to 30° to 40°C in a  
25 constant-temperature water bath to remove the organic solvent. The solid matter was collected by filtration through a glass filter, rinsed five times with 10 ml of distilled water, spread over a glass dish, dried under reduced pressure (0.2 Torr) at 40°C for 24 hours, and  
30 sieved through a screen of 100 mesh\* to produce microcapsules of TAP-144.

The microcapsules showed a remaining amount of methylene chloride of not more than 1 ppm.

Note: \* "mesh" is as specified by the Japanese  
35 Industrial Standard.

CLAIMS

1. A method for producing a formed product of a high molecular compound which is substantially free of halogenated solvent, which method comprises mixing:
  - (a) a high molecular compound having a glass transition temperature of not more than about 60°C,
  - (b) a halogenated solvent and
  - (c) an organic solvent other than the halogenated solvent which is compatible with the halogenated solvent and is less apt to dissolve the high molecular compound, to prepare a solution of the high molecular compound and then maintaining the solution at a temperature around the boiling point of the halogenated solvent so as to remove the halogenated solvent.
2. A method as claimed in Claim 1, wherein the high molecular compound is an ester type high molecular compound.
3. A method as claimed in Claim 2, wherein the ester type high molecular compound is polylactic acid.
4. A method as claimed in Claim 2, wherein the ester type high molecular compound is a copolymer of lactic acid and glycolic acid.
5. A method as claimed in any of Claims 1 to 4, wherein the halogenated solvent is an halogenated methane.
6. A method as claimed in Claim 5, wherein the halogenated methane is methylene chloride.
7. A method as claimed in any of Claims 1 to 6, wherein the organic solvent is an hydrocarbon.
8. A method as claimed in Claim 7, wherein the hydrocarbon is n-pentane.
9. A method as claimed in any of Claims 1 to 8 wherein the temperature around the boiling point of the halogenated solvent is a temperature within 20°C of the boiling point of the halogenated solvent.



European Patent  
Office

## EUROPEAN SEARCH REPORT

0178824  
Application number

EP 85 30 7083

| DOCUMENTS CONSIDERED TO BE RELEVANT   |   |  |   |
|---|---|--|---|
| Category  | Citation of document with indication, where appropriate, of relevant passages | Relevant to claim                              | CLASSIFICATION OF THE APPLICATION (Int. Cl.4)               |
| A   | EP-A-0 054 627 (FÜSSENER TEXTIL AG)<br>* Claims 1-4; page 5, lines 8-14<br>*  | 1  | C 08 J 3/14<br>C 08 G 63/74<br>C 10 G 73/42<br>C 11 B 11/00 |
| A   | GB-A- 743 052 (PROMOTORS & INVENTORS LTD.)<br>* Claims 1,3-7 *                | 1  |   |
| A   | EP-A-0 094 536 (AKZO GmbH)<br>* Claims 1-7 *                                  | 1-4  |   |
|   |   |  | TECHNICAL FIELDS SEARCHED (Int. Cl.4)                       |
|   |   |  | C 08 G<br>C 08 J<br>C 11 B<br>C 10 G                        |
| The present search report has been drawn up for all claims  |   |  |   |
| Place of search<br>THE HAGUE  |   | Date of completion of the search<br>08-01-1986 | Examiner<br>DECOCKER L.                                     |
| <b>CATEGORY OF CITED DOCUMENTS</b><br>X : particularly relevant if taken alone<br>Y : particularly relevant if combined with another document of the same category<br>A : technological background<br>O : non-written disclosure<br>P : intermediate document<br>T : theory or principle underlying the invention<br>E : earlier patent document, but published on, or after the filing date<br>D : document cited in the application<br>L : document cited for other reasons<br>S : member of the same patent family, corresponding document |   |  |   |